

Evaluation of “P-recovery” toolkit based on Wetsus research



PHOSPHORUS recovery

we tested the feasibility of recovery phosphorus
from synthetic urine as



SYNTHETIC URINE RECIPE

compound	FORMULA	MM	Concentration g/L	g for 500mL
Urea	$\text{CH}_4\text{N}_2\text{O}$	60,062	25	12,5
Sodium chloride	NaCl	58,44	4,6	2,3
Potassium dihydrogen phosphate	KH_2PO_4	136,086	4,2	2,1
Potassium chloride	KCl	75	1,6	1,3

PHOSPHORUS recovery

- When enzyme urease is added to the urine causes the conversion of urea into ammonia with increasing of the pH of the solution



enzyme urease from soybeans

- Hydrated soybeans were mashed with a pestle and a mortar
- added to the solution
- The pH of the solution become more than 10



Struvite precipitation

- The solution was filtered and MgSO_4 was added to the solution



- the suspension obtained was filtered and air dried

Struvite as fertilizer



1 week



2 week



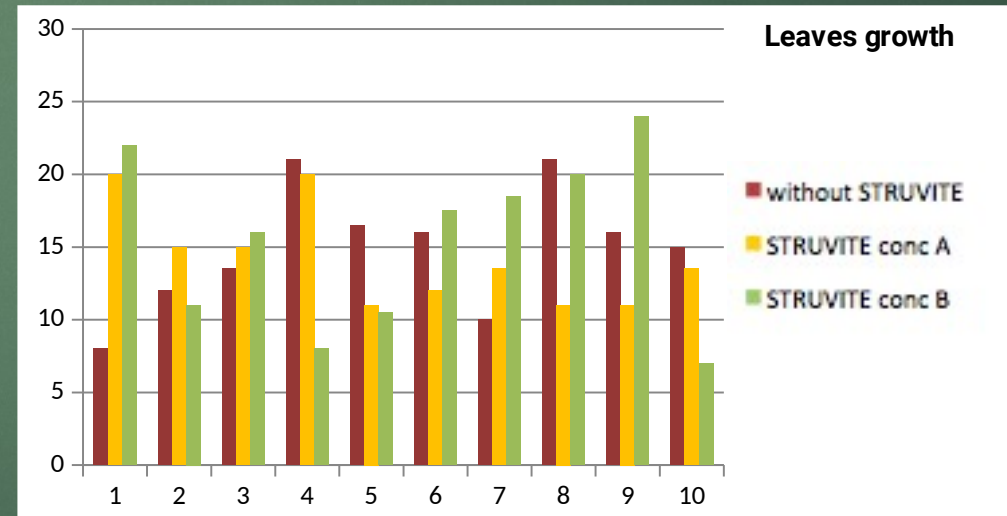
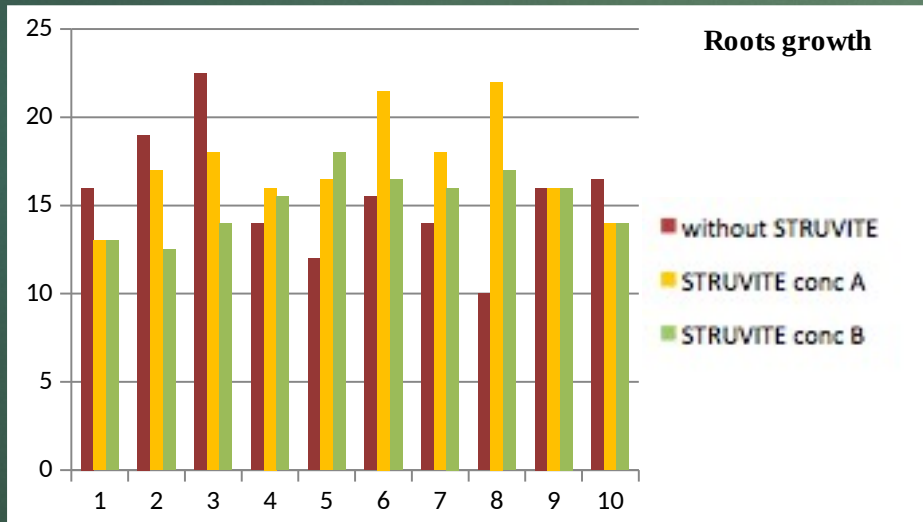
3 week



4 week

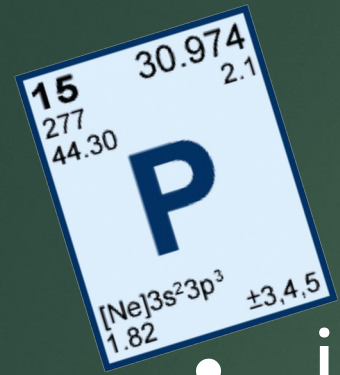
Struvite as fertilizer: results

- We studied the growth of roots and the growth of leaves with two concentration of struvite and without struvite as control
- there isn't a significative difference in the plants growth but plants with struvite grew a little more



What we have learnt:

- waste products contain valuable and useful compounds
- importance of nutrients recovery from urine: phosphate and nitrogen



PHOSPHORUS

- is an indispensable element for plants and animal life
- is a part of the DNA and RNA
- is important for energy transfer in cells as part of ATP (adenosine triphosphate)
- is found in Nature as phosphate in mineral rocks

PHOSPHORUS

- in the 2014 phosphate rocks were included in Critical Raw Materials by EC
- stocks will run out in about 100 years
- recovering from mines is finite
- demand will increase by the strong growth of the world population



Conclusion

Increased efficiency in nutrient capture

- will reduce nutrient load in the environment,
- reduce the costs of wastewater treatment and
- promote recycling of nutrients.



Thanks!

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