



# Process chains in recycling

A brief story of an old TV

## Collection

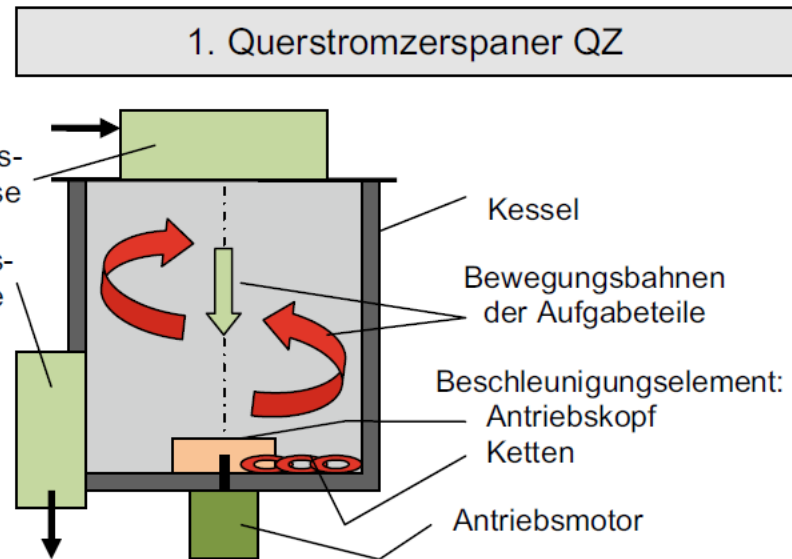
- Old electronic devices are collected locally
- Unlike household waste materials they are not picked up directly at home
- Collection systems provide the first chance to separate waste streams



## Dismantling and shredding

- Dismantling is always the first step in processing larger pieces
  - Larger building blocks can be separated
  - Assortation for the steps to follow
  - Exclusion of toxins etc. possible
- The large pieces from the dismantling process are broken down to a smaller size by shredding

## Shredding



## Keystep: Separation of different grain sizes (sieving)

- Standardization of grain size is of crucial importance for the processes to follow

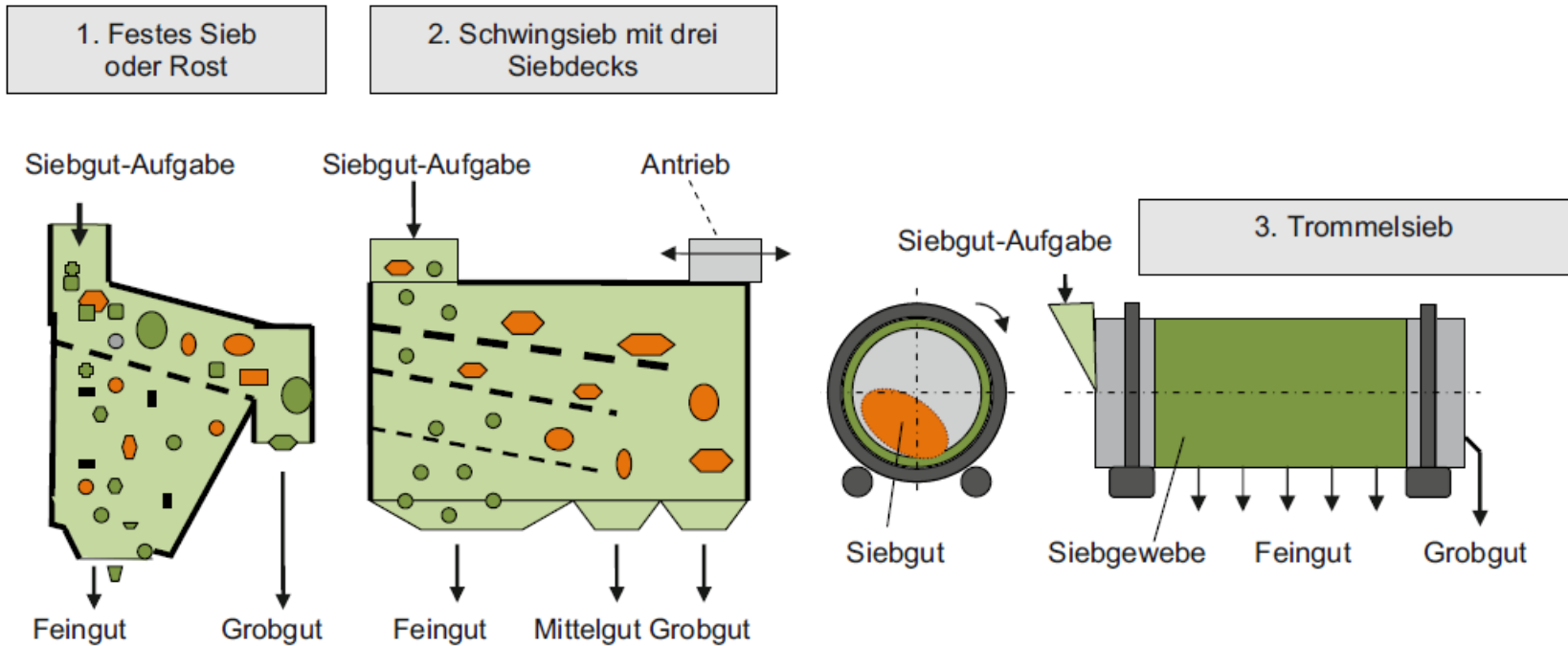


Gemisch

Grobfraktion

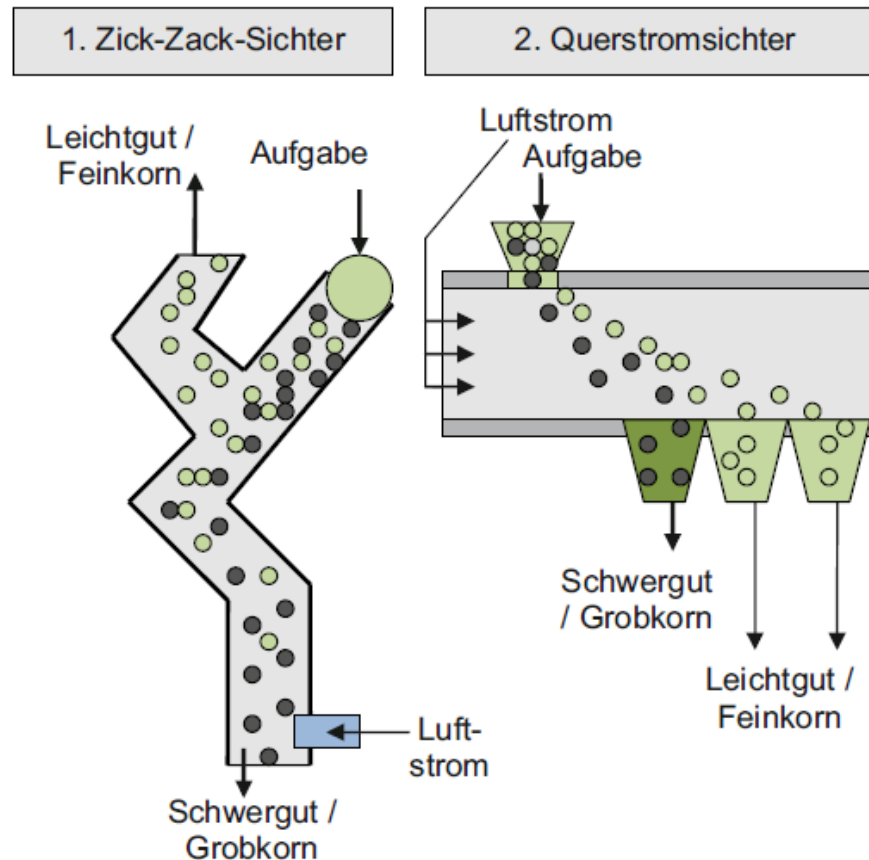
Feinfraktion

## Types of sieves



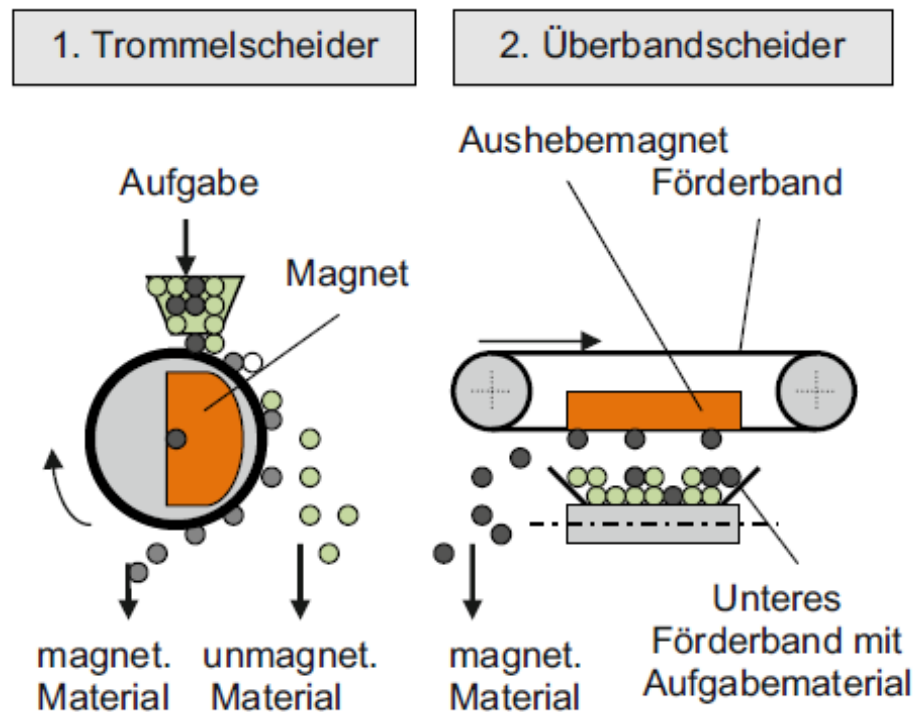
## Separation in the gravitation field

- Particles are divided by their sinking/floating behaviour in fluids (air, water, etc.)
- [Film](#)



## Separation in magnetic fields

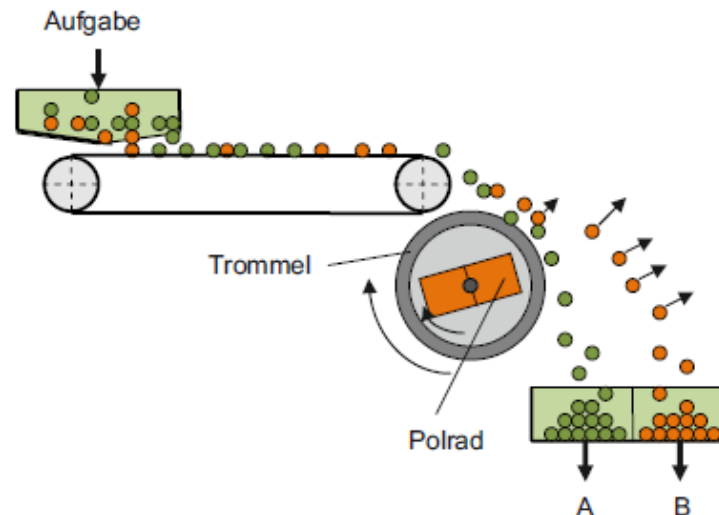
- A magnet is used to pick ferromagnetic materials from the waste stream
- i.e. low alloyed steel, Cr-steel, Ni-Cu-alloys (> 65% Ni), cast iron
- [Film](#)





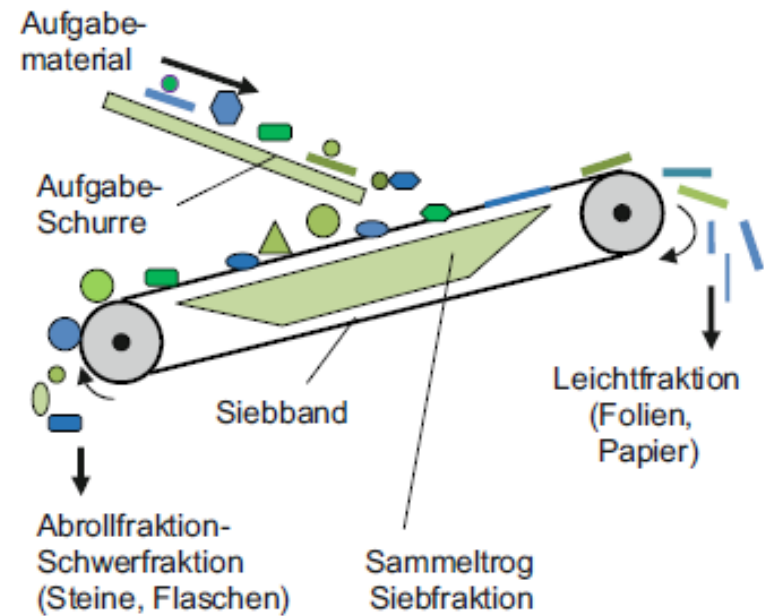
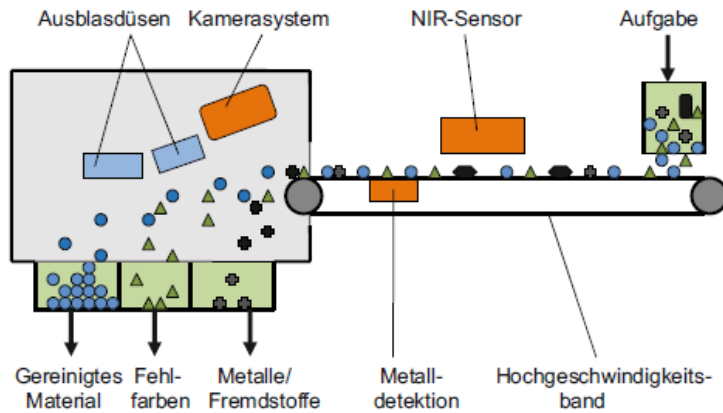
## Separation in electromagnetic fields

- Eddy-current separators separate by the conductivity/density ratio
- Basically conductors are separated from non-conductors
- Grain sizes > 5mm needed
- Absence of ferromagnetic materials is necessary
- Electromagnetic field is induced in conductive grains
  - Repulsion of this material
- Film



A: Leitfähigkeit / Dichte gering  
 B: Leitfähigkeit / Dichte hoch

## Other separation techniques



## What happened until now?

- Collection systems provide a first separation step
- Dismantling and extraction of toxic materials are important for further processing
- Shredders and mills are used to break different materials apart
- The particles have to be assorted by their grain size (sieving) to make further separation techniques possible
- Separation steps use differences in
  - Density
  - Magnetism
  - Conductivity
  - Optical properties
  - Geometrical properties

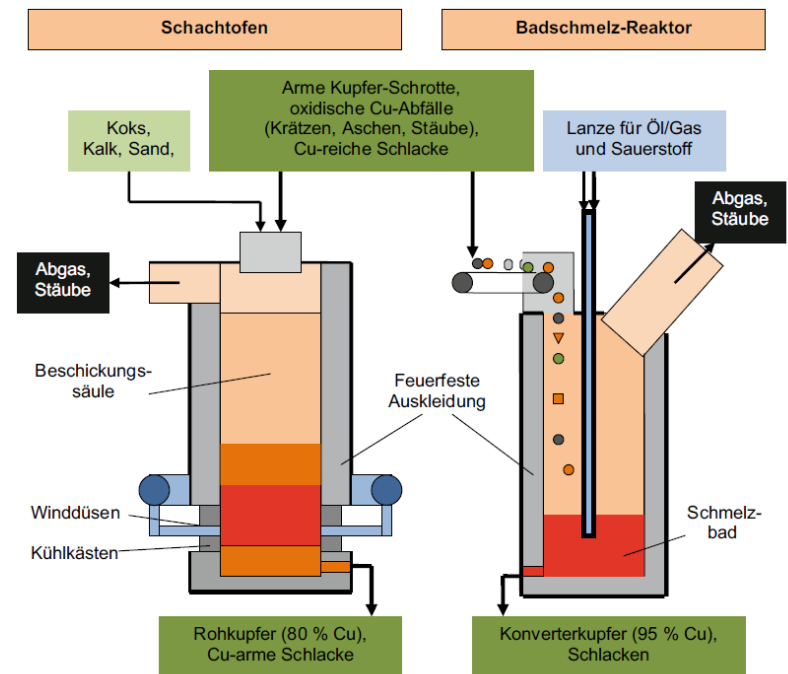
## What's next?

- After sorting the waste particles into fractions of higher content of different kind of metals
  - Pyrometallurgical processes or
  - Hydrometallurgical processes follow



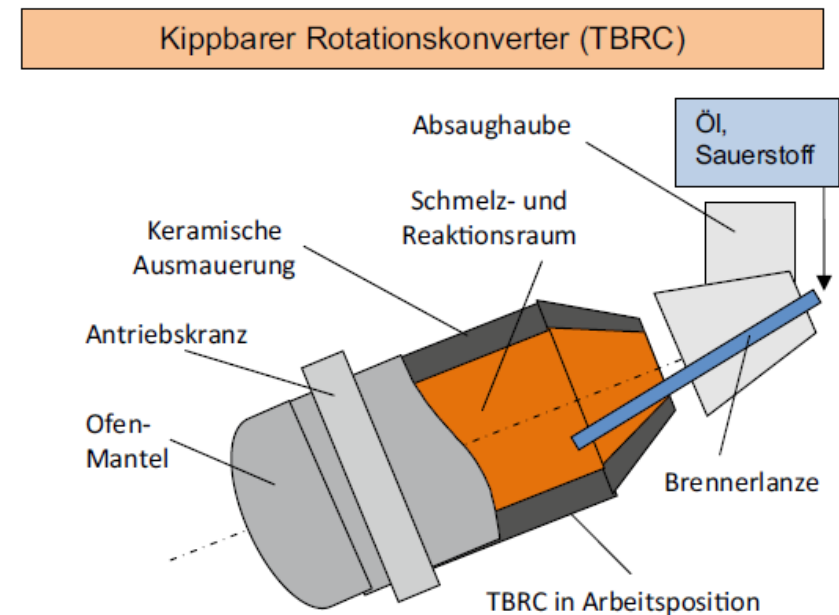
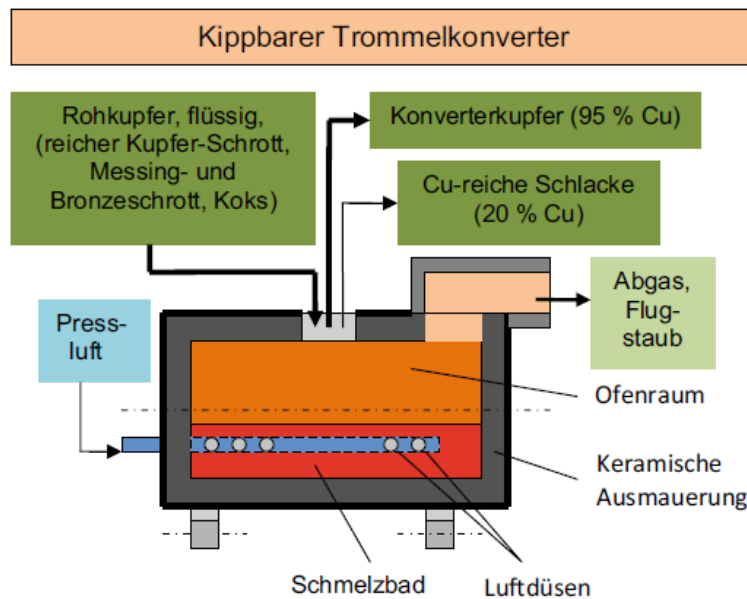
## Further treatment of copper – pyrometallurgical approach

- Copper containing waste is introduced into a furnace to
  - Homogenization of the metal phase
  - Concentration of valuable metals in the molten phase
  - Formation of cinders (FeO-silicate)
  - Collection of dust with ZnO, PbO
  - Treatment of exhaust gases (CO<sub>2</sub>, hydrocarbons, SO<sub>2</sub>, HCl)
- Copper from this processes consists typically of **75% Cu**, 6% Sn, 5% Fe, 3% Ni, 5% Zn, 4% Pb
- Electro-furnaces can be used as well



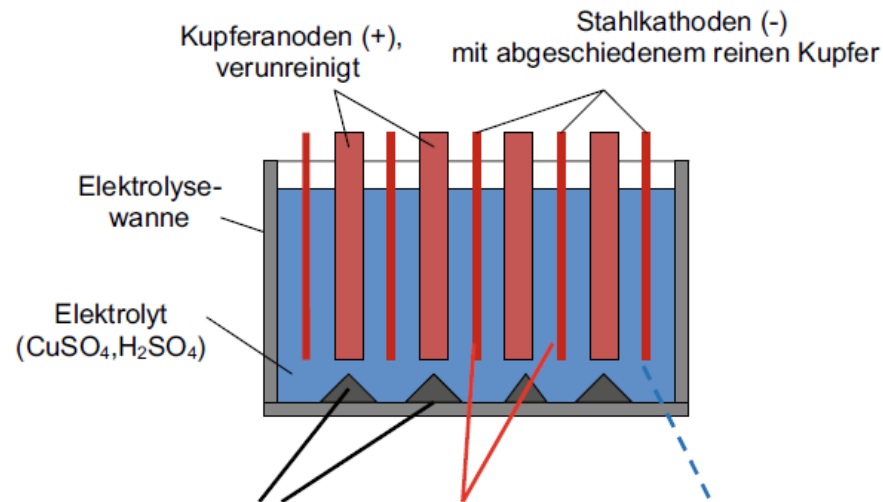
## Converter process

- Selective oxidation
- Extraction of Sn, Pb, Fe, Zn, Al
- Ni and RE-metals remain in the molten copper
- **95 – 98 % copper** content



## Electrolytical refining

- Requires 99 – 99,5 % copper content
- Outcome: **99,98% Cu**



	Anodenschlamm			Kathoden-Kupfer	Elektrolyt (CuSO <sub>4</sub> , H <sub>2</sub> SO <sub>4</sub> )					
Normal-Potential (V):	Pt +1,6	Au +1,4	Ag +0,81	Cu +0,35	H 0	Pb -0,13	Sn -0,14	Ni -0,25	Fe -0,44	Zn -0,76
Elektrochemischer Effekt:	Elektrochemisch unlöslich.			Abscheidung an der Kathode, hochrein, kompakt.	Elektrochemisch löslich. Keine Abscheidung vor dem Kupfer und dadurch Anreicherung dieser Metalle im Elektrolyt.					

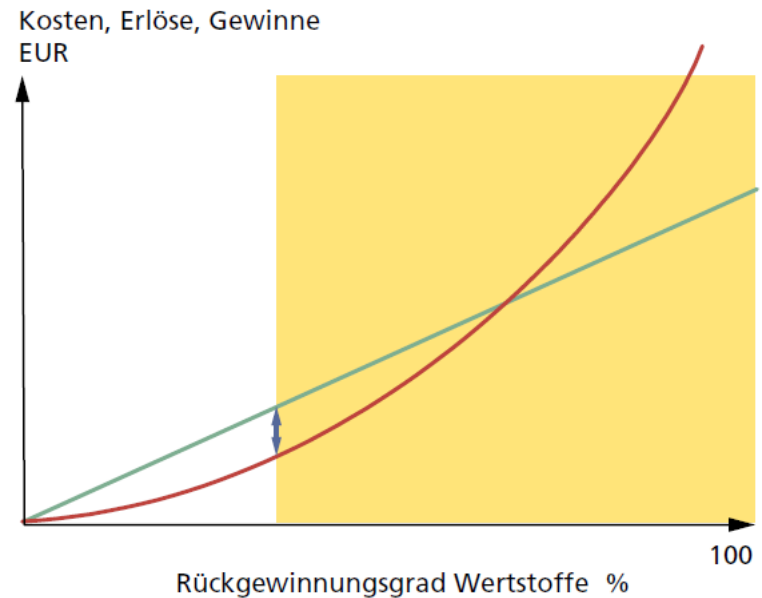
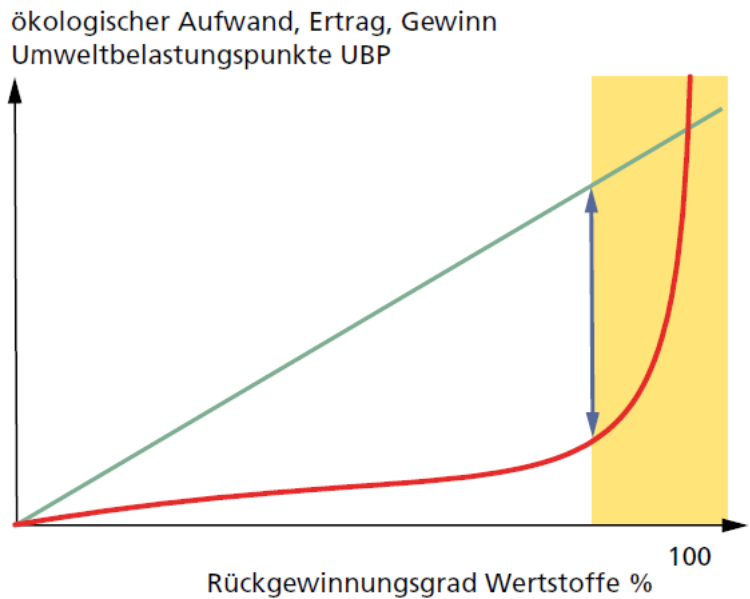
## Further treatment of copper – hydrometallurgical approach

- Only useful if copper content is rather low
- Leaching
  - with ammoniumcarbonate/ammonia (Cu and Ni)
  - with iron(III)sulphate (not very selective)
  - with sulfuric acid (not very selective)
- Copper recovery from aqueous solutions
  - Electrolytic precipitation (reduction) – requires  $> 15$  g/L Cu
  - Cementation with metallic iron (not very selective, 60 – 90% Cu)
  - Thermal decomposition of Cu-tetramincomplex (regeneration of ammonia)
  - Precipitation as hydroxide or sulfide (both less favoured)



## Limitations of recycling

- Economical, ecological, thermodynamical



— maximaler ökologischer Gewinn    
 — ökologischer Aufwand    
 — ökologischer Erlös

$$\text{Gewinn} = \text{Erlös} - \text{Aufwand}$$

— maximaler wirtschaftlicher Gewinn    
 — Kosten    
 — Erlös

$$\text{Gewinn} = \text{Erlös} - \text{Kosten}$$

Thank you for your find attention



## Treatment of cables and wires

